

Phosphorus and Potash Availability in the Salt Affected Soil of Bhal Region (Gujarat) as Influenced by Natural Drying, Levels of Phosphorus Sources and FYM

V.S. Patel and J.C. Patel

Department of Agricultural Chemistry and Soil Science,
Gujarat Agricultural University, Anand 388 110, India

Abstract : Incubation of a 'Bhal' soil with or without farm yard manure (FYM) and phosphorus from different sources showed that application of FYM markedly decreased the soil pH and significantly enhanced the P and K availability in the soil. The P and K availability of soil reduced significantly when moisture was depleted from 40 to 10%. Again on rewetting the soil, nutrients availability was restored. Application of diammonium phosphate significantly increased the Olsen's P in the soil as compared to that with monocalcium phosphate.

Key words : P K availability, salt-affected soil, moisture regime, P sources and levels, FYM.

In Gujarat state, Bhal region is known for growing durum wheat. Most of the soils of this region are clay to clay loam in texture. Average rainfall of this region is 600 mm. But, soils get flooded even with a small amount of rainfall because of flat topography and poor water transmission characteristics (Patel *et al.*, 1993). Durum wheat (*rabi*) is grown in this area with the help of residual moisture (30 to 40%) conserved in sub-soil during monsoon. Farmers apply only nitrogen in the form of urea along with the seed. Though most of the soils are deficient in phosphorus and marginal in zinc availability, deficiency symptoms are not noticed on wheat crop (Motiramani *et al.*, 1990).

In this context, an incubation study was carried out in the laboratory to understand the changes in nutrient availability of the soil at different moisture regimes simulated by natural drying of the saturated soil with or without addition of FYM and P from different sources, and rewetting.

Materials and Methods

A bulk soil (clay loam) sample (0 to 20 cm) was collected from a farmer's field (Arnej, Bhal area), air dried and powdered (2 mm). The physico-chemical characteristics of the soil were $pH_{2.5}$ 7.76, $EC_{2.5}$ 0.9 $dS\ m^{-1}$, CEC 33.27 $C\ mol\ kg^{-1}$, ESP 19.1, maximum water holding capacity 50%, organic matter 0.65%, $CaCO_3$ 9.34%, total N 0.04%, Olsen's P 9.41 $kg\ ha^{-1}$ and ammonium acetate extractable K 255 $kg\ ha^{-1}$ (Jackson, 1973).

Plastic containers were filled with 150 g of treated soil. Various treatments consisted of three levels of P (0, 25 and 50 $kg\ P_2O_5\ ha^{-1}$) from two sources (AR Diammonium phosphate and monocalcium phosphate), two levels of FYM (0 and 13.5 $t\ ha^{-1}$) and five moisture regimes (M₄₀, M₃₀, M₂₀, M₁₀, MR₄₀). Soils under each treatment were incubated for three weeks under flooding condition and then allowed to dry naturally to desired levels of moisture. It took 30, 60, 85, 115 days drying after three weeks flooding to obtain 40, 30, 20 and 10% moisture in the soil, respectively.

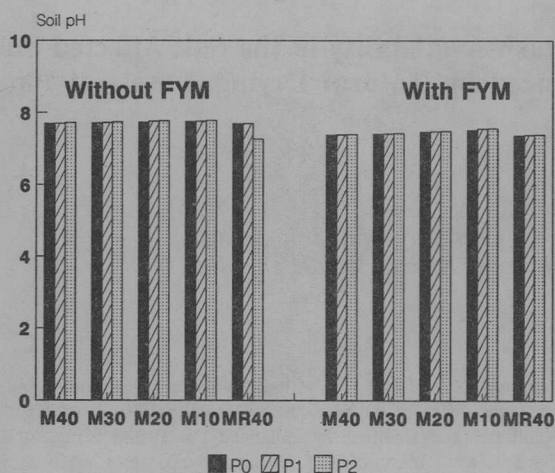


Fig. 1. Average effect of moisture regimes and FYM on soil pH.

Such wetting-drying cycle normally occurs in Bhal area. In case of MR₄₀ treatment, after the completion of first cycle of wetting-drying, the above process was repeated to bring the soil to 40% moisture. Each treatment was replicated twice. The wet soil samples at different moisture levels/periods were immediately analyzed for moisture content, pH, Olsen's P and ammonium acetate extractable K using standard methods. Per cent P fixed was also computed by subtracting P increased over P₀ from the applied phosphorus. The data were statistically analysed using completely randomized design.

Results and Discussion

Soil reaction

The pH values increased when the moisture reduced from 40 to 10%. Application of FYM markedly decreased the soil pH from 7.73 to 7.44 (Fig. 1). The different treatment combinations of moisture regimes with amounts of P did not show any marked effect on soil pH. The results clearly indicated that moisture and FYM had a marked role in decreasing the soil pH. The reduction in soil pH at higher moisture level could be attributed to the depletion of oxygen, ac-

cumulation of CO₂ and building up of the partial pressure of CO₂ in alkaline and calcareous soil (Ponnamperuma, 1977). The decrease in soil pH due to FYM application could be attributed to the production of H₂CO₃ and other organic acids during microbial decomposition of organic matter (Singh and Lal, 1976).

Olsen's P content

Moisture : The data in Table 1 show that Olsen's P was significantly decreased when the soil moisture was depleted from 40 to 10%, the decrease was as high as 73%. The decrease in P at M₂₀ and M₃₀ was 26 and 51%, respectively. Also the effect of M₄₀ and MR₄₀ on the available P content in the soil was similar. Thus, the availability of P was restored on rewetting the soil.

The significantly higher available P content with the increase in the level of P application was observed at each moisture regime. However, with the depletion in moisture at each level of P, the availability of P decreased. The maximum P availability was observed under P₂ level at each moisture level. The decreased availability of P with the depletion in moisture could be attributed to the oxidation

Table 1. Effect of moisture regimes, amount and sources of P and FYM on estimates of plant-available P* (ppm) in soil

	M40	M30	M20	M10	MR40	Mean
Phosphorus						
P ₀	6.54	5.35	4.51	3.92	6.58	5.38
P ₁	10.23	7.94	6.69	6.01	9.18	8.01
P ₂	12.26	9.72	8.01	6.88	13.12	10.00
Source						
S ₁	8.42	6.58	5.51	5.06	9.34	6.98
S ₂	10.94	8.76	7.12	6.13	9.92	8.57
FYM						
F ₀	5.96	4.93	4.02	3.55	5.48	4.79
F ₁	13.39	10.41	8.79	7.66	13.78	10.81

* Olsen's extractable.

CD 5%, M = 0.201, P = 0.156, F/S = 0.127, M x P = 0.349, M x S/F = 0.285.

of more soluble ferrous phosphate to insoluble ferric phosphate and also due to precipitation of calcium phosphate under reduced concentration of CO₂ (Ponnamperuma, 1967, 1972). Such a change is supported by increase in pH (Fig. 1) and fixation of P with depletion in soil moisture content (Fig. 2).

P source : There was significant increase in Olsen's P with the increase in level of P application. When P was applied through DAP (S₂), the available P in the soil was significantly higher (22%) than in the case of MCP (S₁) due to less P fixation under

former one (Fig. 2). The significantly more available P content was observed with P₂S₂ treatment combination over rest of the treatments.

Farm yard manure : Application of FYM (13.5 t ha⁻¹) significantly increased (more than double) the Olsen's P in the soil at all the levels of moisture, with maximum P availability under M₄₀F₁ or MR₄₀F₁ treatment combinations. The data also revealed that the Olsen's P content was significantly increased (59%) even with only FYM application in absence of P. However, the increase was more than

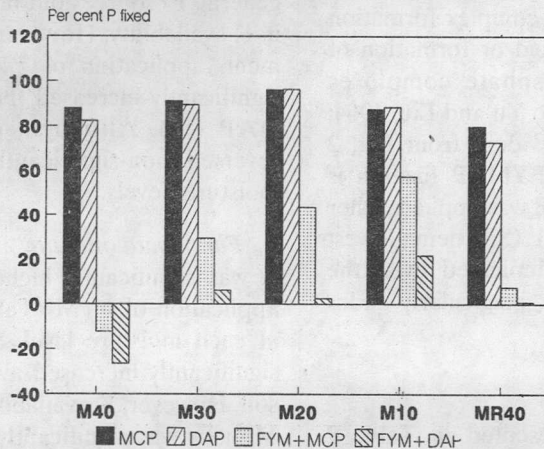


Fig. 2. Per cent P fixed (of the applied) as influenced by moisture regimes and FYM.

Table 2. Effect of moisture regimes, amount and sources of P and FYM on estimates of plant available K (kg ha⁻¹) in soil

	M ₄₀	M ₃₀	M ₂₀	M ₁₀	MR ₄₀	Mean
Phosphorus						
P ₀	332	316	275	249	331	301
P ₁	340	324	284	258	337	307
P ₂	348	330	288	255	344	313
Source						
S ₁	341	322	281	250	340	307
S ₂	339	325	284	254	335	307
FYM						
F ₀	320	322	276	239	311	291
F ₁	361	325	289	265	363	324

CD 5%, M = 8.0, P = 2.3, F = 1.9, M x F/S = 4.3.

double with P₁F₁ and about three times more with P₂F₁ treatment combination over that of P₁F₀ and P₂F₀, respectively.

Furthermore, Olsen's P in the soil was significantly higher (24%) when P was applied through DAP (S₂) than through MCP (S₁) in the presence of FYM.

The increased availability of P on FYM application may be due to either directly by release of P from iron and/or aluminium phosphate or by influencing: (i) the decrease in redox potential, (ii) pH changes (Fig. 1), and (iii) mobilization of P by complex formation between Fe and Fulvic acid or formation of soluble fluvic iron phosphate complexes (Bradley and Sieling, 1953; Yu and Liu, 1964; Sinha, 1972). It is also evident from Fig. 2 that in the presence of FYM, P fixation is drastically reduced when P was applied either through DAP (S₂) or MCP (S₁), being lowest under DAP. Data also indicated even the release of P due to FYM, particularly under higher moisture levels.

Potassium availability

Moisture : Data presented in Table 2 clearly indicated the significant decrease in ammonium acetate extractable K content in

soil with the depletion in moisture from 40 to 10% (M₄₀ and MR₄₀ were at par). It could be attributed to fixation of K in the lattice of the mountmorillonitic Bhal-soil (Nedeco, 1968) during drying period and again release of K on rewetting (Savant *et al.*, 1981; Grewal and Kanwar, 1967).

P source : The data revealed that application of P at both the levels significantly increased the K availability in the soil. However, the increase was very meagre, being 2.2 and 4.1% with P₁ and P₂ levels, respectively. In general, P sources did not differ significantly in K availability. However, under MR₄₀ treatment, application of P through MCP (S₁) significantly increased the K availability over DAP (S₂). Although, the behaviour was reversed (non-significantly) at 30, 20 and 10% moisture levels.

Farm yard manure : The availability of K was significantly higher (11.3%) with the application of FYM (Table 2). Furthermore, at each moisture level, application of FYM significantly increased available K content in soil. However, K availability under M₄₀F₁ and MR₄₀F₁ was significantly higher over rest of the treatment combinations. The K availability was also significantly increased with FYM

application at all P levels, being maximum in the case of P₂F₁. Similarly, significantly more K availability was observed when DAP (S₂) was applied along with FYM. The beneficial effect of FYM in enhancing the K availability may be due to addition of K (5279 ppm) through FYM. Brar and Sekhon (1987) also noted positive significant correlation between organic carbon and exchangeable K in five bench mark soil series of northern India.

The results of the present investigation suggest that preapplication of P through DAP, along with 13.5 t ha⁻¹ FYM and sowing at 40% soil moisture level to P deficient calcareous soil of Bhal area, could favourably influence the P and K nutrition of durum wheat.

References

- Bradley, D.B. and Sieling, D.H. 1953. Effect of organic anions and sugars on phosphate precipitation by iron and aluminium as influenced by pH. *Soil Science* 46: 175-179.
- Brar, M.S. and Sekhon, G.S. 1987. Vertical distribution of potassium in five benchmark soil series in northern India. *Journal of the Indian Society of Soil Science* 35: 732-735.
- Grewal, J.S. and Kanwar, J.S. 1967. Potassium fixation in some soils of Punjab, Haryana and Himachal Pradesh. *Journal of the Indian Society of Soil Science* 15: 237-244.
- Jackson, M.L. 1973. *Soil Chemical Analysis*. Prentice Hall of India Pvt. Ltd., New Delhi.
- Motiramani, D.P., Patel, J.C., Giri, J.D., Raman, S. and Agrawal, J.H. 1990. Evaluation of phosphorimeter for estimation of phosphorus availability of Bhal and coastal soils in Gujarat state. *Gujarat Agriculture University Research Journal* 16: 53-56.
- Nedeco, 1968. *Reports on Bhal reclamation scheme in state of Gujarat, India. Vol. I to IV*, Grontmij N.V. de Biltz, the Hague, Netherlands.
- Patel, J.C., Patel, B.D. and Giri, J.D. 1993. Genesis characterization and classification of some salt affected soils of Bhal-Nal area, Gujarat. *Journal of the Indian Society of Coastal Agriculture Research* 11: 1-12.
- Ponnamperuma, F.N. 1967. A theoretical study of aqueous carbonate equilibria. *Soil Science* 103: 90-100.
- Ponnamperuma, F.N. 1972. The chemistry of submerged soils. *Advances in Agronomy* 24: 29-96.
- Ponnamperuma, F.N. 1977. Physico-chemical properties of submerged soil in relation to fertility. *IRRI Research Paper Series*, No. 5, pp. 3-29.
- Savant, R.D., Kadrekar, S.B. and Dongale, J.H. 1981. Phosphorus and potassium availability as affected by organic matter and different moisture regimes. *Maharashtra Agriculture University Journal* 6: 179-182.
- Singh, A. and Lal, B. 1976. Organic matter in soil and its maintenance. In *Soil Fertility, Theory and Practice* (Ed. J.S. Kanwar), pp. 128-155. ICAR, New Delhi.
- Sinha, M.K. 1972. Organo-metallic phosphates IV. The solvent action of fulvic acids on insoluble phosphates. *Plant and Soil* 37: 457-467.
- Yu, T. J. and Liu, C.K. 1964. Oxidation reduction processes in paddy soils and their relation to rice growth. *Acta Pedol. sin.* 12: 380-389.